

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE  
BEFORE THE BOARD OF PATENT APPEALS AND INTERFERENCES

In re application of	Appeal No.
Harald KRAUS et al.	Conf. 2464
Application No. 10/588,766	Group 1792
Filed August 8, 2006	Examiner Roberts Culbert
METHOD FOR SELECTIVE ETCHING	

**REPLY BRIEF**

MAY IT PLEASE YOUR HONORS:

May 10, 2010

The Examiner's Answer of March 9, 2010 responded to the two erroneous interpretations discussed in the Appeal Brief. The Examiners remarks regarding these erroneous interpretations are discussed below.

**1. CHRISTENSON does not suggest an etchant flow velocity parallel to the substrate surface that could be optimized.**

The Examiner did not find the argument "CHRISTENSON does not suggest an etchant flow velocity parallel to the substrate" to be persuasive because the Examiner relied on the combination of CHRISTENSON and TANAKA.

Examiner states that "Christenson teaches a flow of 0.5 to 2 lpm for single wafer spray processors". Appellant has never disagreed with this fact.

Appellant also has never disagreed that TANAKA teaches a free-beam, which leads the respective velocity  $v$  parallel to the substrate's surface of minimum 0.1 m/s.

However, the Examiner (at item b. on page 6 of the Examiner's Answer) alleges that the volume flow of "0.5 to 2 lpm" of CHRISTENSON "... is sufficient to generate a mean velocity  $v$  parallel to the substrate's surface of minimum 0.1 m/s ...". There is no finding of fact to support this conclusion.

That is, the Examiner does not explain where this velocity may be found in CHRISTENSON, or how this velocity (in meters per second) was determined based on the volumetric flow rate (liters per minute) taught by CHRISTENSON.

Consequently, this allegation remains a mere postulation.

**2. CHRISTENSON teaches away from using either  $\text{HfO}_2$  or  $\text{ZrO}_2$ .**

The Examiner did not find the argument "CHRISTENSON teaches away from etching using  $\text{HfO}_2$  and  $\text{ZrO}_2$ " to be persuasive because the Examiner selected that feature not from CHRISTENSON but rather from BUCHANAN.

Appellant maintains that CHRISTENSEN explicitly teaches away from using the claimed  $\text{HfO}_2$  or  $\text{ZrO}_2$  because "a material comprising only one elemental constituent other an oxygen (e.g.,

HfO<sub>2</sub> or ZrO<sub>2</sub>), are highly resistant to dilute etchants" (paragraph 22 of CHRISTENSON).

Moreover, while CHRISTENSON teaches HfSiO<sub>2</sub> may be substituted for HfO<sub>2</sub>, as identified by the Examiner, HfSiO<sub>2</sub> is a completely different compound from HfO<sub>2</sub>. This material is not only different in the chemical composition, but also in its crystals. That is, CHRISTENSON requires an additional elemental constituent, i.e., Si. The interpretation that HfO<sub>2</sub> is understood to be "substituted HfO<sub>2</sub>" is simply chemically incorrect.

The ion bombardment that BUCHANAN teaches does not turn HfO<sub>2</sub> into HfSiO<sub>2</sub> so it does not turn HfO<sub>2</sub> into a different chemical substance. The ion bombardment rather brings cracks into the crystal structure of HfO<sub>2</sub> and thus enables HfO<sub>2</sub> to be etched by wet chemicals.

Thus, the fact remains that the dilute solutions utilized by CHRISTENSON are incompatible with materials including HfO<sub>2</sub>, and, as explained in the Brief, HfO<sub>2</sub> would have rendered the materials of CHRISTENSON unsatisfactory for use with the dilute solutions of CHRISTENSON.

### **3. Conclusion**

The above discussion is believed to underscore that the remaining rejections of the claims on appeal are improper and should be reversed. Such action is accordingly respectfully requested.

Respectfully submitted,

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